

Interacting effects of an increase of pCO₂ and temperature on photosynthesis and calcification in a scleractinian coral

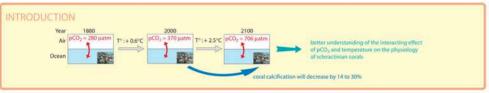
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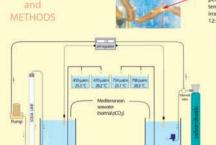
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This poster is dedicated to Samantha Romaine-Lioud.





Stylophora pistillata (40 nubbins suspended on nylon strings)

temperature = 25°C

irradiance = 380 µmol m⁻² s⁻¹

Experimental set up

- Four culture conditions * 450 µatm-25.3°C ("normal pCO₂, normal temperature")
- * 470 µatm-28.2°C ("normal pCO₂, high temperature")
- * 734 yearm-25 1°C ("high oCO» normal temperature")
- * 798 µatm-28.3°C ("high pCO₂, high temperature").

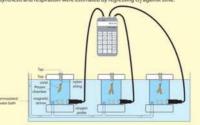
All colonies were initially kept for 2 weeks under "normal pCO2, normal temperature". Then, 10 colonies were randomly dispatched in each of the four tanks and the experiment ran for 5 more

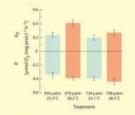
Control of seawater pCO2

Seawater pCO₃ was adjusted prior to the transfer into aquaria using a pH controller (R305, Consort Inc.) connected to pH electrodes (Orion, model 8102SC) as described by Leclercq et al. (2000). pH modifications were achieved by bubbling seawater with either pure CO2 (to increase pCO2) or with CO-free air (to decrease pCO-).

Since pCO₂ was controlled by injecting gases, total alkalinity was not affected with the result that the changes of the carbonate chemistry were properly mimicking the changes predicted to occur during the next decades.

Net photosynthesis (P_n) and dark respiration (R) were measured on 3 colonies taken in each of the four tanks. 3 Perspex chambers (240 ml) filled with the seawater used in each treatment were used simultaneously in a thermostated water bath. The incubation medium was continuously agitated. Dissolved O₂ was measured using a Ponselle polarographic electrode and monitored every 1 min using a data-logger (LI-1000, Li-COR). Rates of net photosynthesis and respiration were estimated by regre





Pn of each colony measured during the 5 weeks subsequent to the perturbation did not vary with time (repeated measures ANOVA, P = 0.15).

P_{ri} was affected by temperature (ANOVA, P = 0.0005) and pCO2 (ANOVA, P = 0.009).

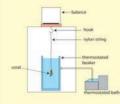
Respiration was not affected by temperature (ANOVA. P = 0.12), nor by pCO₂ (ANOVA, P = 0.11).

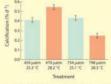
The increase of photosynthesis with increasing temperature under normal pCO2 is in agreement with previous studies performed on corals (Coles & Jokiel, 1978; Kajiwara et al., 1995).

Elevated pCO₂ did not stimulate photosynthesis, which even slightly decreased. Langdon et al. (2003) also showed that net community production did not change in response to elevated pCO₃.

Corals are known to rely on bicarbonate for photosynthesis (Goiran et al., 1996). The increase in pCO2 results in higher concentrations of dissolved CO₂ and bicarbonate, but the increase of the bicarbonate reservoir in which corals pump carbon for photosynthesis is likely too small (9 to 10%) to lead to a measurable increase of photosynthesis.

The skeletal dry weight was measured every week by weighing each colony using the buoyant weight technique (Jokiel et al., 1978; Davies, 1989).





The rate of calcification of each colony did not vary significantly after the perturbation (repeated measures ANOVA, P = 0.3).

The calcification rate was significantly affected by the treatment (ANOVA, P < 0.0001 for pCO2 and P = 0.3 for temperature). The significant interaction (P < 0.001) between pCOs and temperature demonstrates that there was a response to a change in temperature but that it differs depending on the level of pCO2.

The rate of calcification declined immediately after the rise in pCO2 and did not change afterwards, strating that no acclimation process occurred. This is in agreement with previous studies:

* Marubini & Atkinson (1999) reported that the decrease of calcification in response to changes in

* Langdon et al. (2000) found that the response of the community calcification rate of the Biosphere 2 ocean is not different during short-term (days) and long-term (months) changes in Ω_{arag}.

Calcification of colonies maintained at elevated temperature declined by 50% in response to increased pCO₂. However, calcification was not affected by elevation of pCO₂ in colonies maintained at normal temperature. This is not in agreement with several papers that describe a negative relationship between calcification and CO2, or a positive relationship with the aragonite saturation state (Gattuso et al., 1998; Marubini & Atkinson, 1999; Langdon et al., 2000; Leclercq et al., 2000; Leclercq et al., 2002; Langdon et al., 2003; Marubini et al., 2003). However, in some of these studies, Ω_{arag} has not been changed by manipulating pCO₂ but by changing the Ca²⁺ concentration (Gattuso et al., 1998), or by addition of acid (Marubini & Thake, 1999; Marubini et al., 2003) or sodium bicarbonate (Marubini & Atkinson, 1999). These results demonstrate that pCO₂ and temperature significantly interact to control calcification. The physiological basis of the different response at two temperatures does not result from an indirect effect of emperature on the seawater carbonate chemistry. Indeed, the change of pH and aragonite saturation state due to increased temperature was similar at both pCO2s (ΔpH: -0.02 to -0.03; ΔΩ2srag: 0.18 to 0.25) and approximately 10 times lower than the changes resulting from increased pCO2.

These results are of major interest from a predictive point of view. Several studies investigated the physiological relationship between calcification and pCO2 or the aragonite saturation state. The consensus opinion is that calcification of tropical marine organisms and coral communities will decrease by an average 18-37% (Gattuso et al., 1999) between preindustrial time and the year 2100. However, none of these studies considered the effect of the forecast increase in temperature and its interaction with pCO₂ on photosynthesis and calcification

Our results demonstrate that the rate of calcification could decrease by 50% between the years 2000 and 2100. This temperature effect must be taken into consideration in subsequent investigations of future changes of coral physiology and reef metabolism. The present predictions must be re-evaluated as our results suggest that the decrease in the rate of calcification at the end of the century could be much higher than that forecast due to the synergistic effects of temperature and pCO₂. There is a pressing need to manipulate environmental parameters in concert in order

mine the response of coral calcification to global environmental changes.

Coles SL, Jokiel PL (1978) Synergistic effects of temperature, salinity and light on the hermatypic coral Montipora versucosa, Mar Biol 49, 187-195

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Marubini F, Thake B (1999) Bicarbonate addition promotes coral growth, Limnol Oceanogr 44, 716-720 Muscatine L. Cernichiari E(1969) Assimilation of photosynthetic products of zooxantheliae by a reef coral

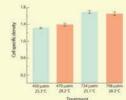
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3 nubbins from each treatment were used to determine the average number of zooyanthellae per animal cell (cell specific density). Corals were crushed with a hammer, placed in a 50 ml flask, and macerated by agitation (Muscatine & Cernichiari, 1969). Approximately 330 host cells from each colony were observed and ranked according to the number of zooxanthellae (from one to eight) that each









The CSD increased under "high pCO2" (ANOVA, P < 0.001) without being affected by the change in temperature (ANOVA, P =

There was a dominance of singlets over doublets or triplets under "normal pCO₂" (70% of singlets and 30% of doublets). The elevated pCO2 with 47% of singlets, 41% of doublets and 11% of animal cells containing more than 2 zooxanthellae.

Under normal pCOs, the CSD was identical at both temperature and equal to 1.4. The same value has been reported in the same species (Muscatine et al., 1998), and indicates that there is a dominance of singlets. This seems to be the standard

condition of the symbiosis.

The CSD increased to 1.7 under elevated pCO2 suggesting a higher rate of algal division compared to the division of animal cells. A change in CSD indicates a disruption of the balance between the growth rate of algal and animal cells.