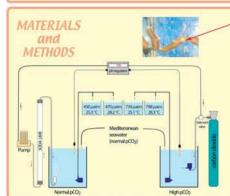


## Interacting effects of pCO2 and temperature on metabolism of a scleractinian coral

S. Reynaud 1, N. Leclercq 1, S. Romaine-Lioud 1, C. Ferrier-Pagès 1, J. Jaubert 1, J.-P. Gattuso 2 and D. Allemand 1

Centre Scientifique de Monaco, Avenue Saint-Martin, MC-98000 Monaco, Principality of Monaco <sup>2</sup> Observatoire Océanologique, Laboratoire d'Océanographie CNRS-UPMC, BP 28, 06234 Villefranche-sur-mer Cedex, France





Stylophora pistillata (40 nubbins suspended on nylon strings)

pCO<sub>2</sub> = 430 µatm temperature = 25°C irradiance = 380 µmol m-2 s-2 weeks (until tissues have entirely recovered the skeleton)

12:12 photoperiod

Experimental set up
Four culture conditions:

\*450 µathr-25.3" ("normal pCO<sub>2</sub>, normal temperature")

\*470 µathr-25.3" ("Crimal pCO<sub>2</sub>, high temperature")

\*734 µathr-25.1" ("high pCO<sub>2</sub>, normal temperature")

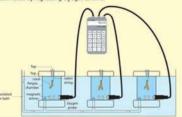
\*798 µathr-28.3" ("high pCO<sub>2</sub>, normal temperature")

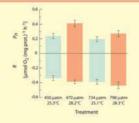
\*798 µathr-28.3" ("high pCO<sub>2</sub>, high temperature")

\*RI colonies were initially kept for 2 weeks under "normal pCO<sub>2</sub>, normal temperature". After, 10 colonies were randomly dispatched in each of the four tanks and the experiment ran for 5 more weeks.

Control of seawater pCO<sub>2</sub>
Seawater pCO, was adjusted prior to the transfer into aquaria using a pH controller (R305, Consort Inc.) connected to pH electrodes (Orion, model 8102SC) as described by Leclercq et al. (2000). pH modifications were achieved by bubbling seawater with either pure CO<sub>2</sub> (to increase pCO<sub>2</sub>) or with CO<sub>2</sub>-free air (to decrease pCO<sub>2</sub>). Since pCO<sub>2</sub> was controlled by injecting gases, total alkalinity was not affected: the changes of the carbonate chemistry were properly mimicking the changes predicted to occur during the next decades.

Net photosynthesis  $(P_n)$  and dark respiration (R) were measured on 3 colonies taken in each of the four tanks. 3 Perspex chambers (240 m) filled with the seawater used in each treatment were used simultaneously in a thermostated water bath. The incubation medium was continuously agitated. Dissolved  $O_2$  was measured using a Ponselle polarographic electrode and monitored every 1 min using a data-logger (L1-1000, L1-COR). Rates of net photosynthesis and respiration were estimated by regressing  $O_2$  against time.





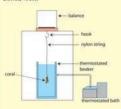
 $P_{\rm ft}$  of each colony measured during the 5 weeks subsequent to the perturbation did not vary with time (repeated measures ANOVA, P = 0.15).  $P_{\rm ft}$  was affected by temperature (ANOVA, P = 0.0005) and pCO<sub>2</sub> (ANOVA, P = 0.009). Respiration was not affected by temperature (ANOVA, P = 0.12), nor by pCO<sub>2</sub> (ANOVA, P = 0.11).

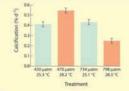
The increase of photosynthesis with increasing temperature under normal pCO<sub>2</sub> is in agreement with previous studies performed on corals (Coles & Jokiel, 1978; Kajiwara et al., 1995).

Elevated pCO<sub>2</sub> did not stimulate photosynthesis, which even slightly decreased. Langdon et al. (2003) also showed that net community production did not change in response to elevated pCO<sub>2</sub>.

Corals are known to rely on bicarbonate for photosynthesis (Goiran et al., 1996). The increase in pCQ<sub>2</sub> results in higher concentrations of dissolved CQ<sub>2</sub> and bicarbonate, but the increase of the bicarbonate reservoir in which corals pump carbon for photosynthesis is likely to small (9 to 10%) to lead to a measurable increase of photosynthesis.

The skeletal dry weight was measured every week by weighing each colony using the buoyant weight technique (Jokiel et al., 1978; Davies, 1989).





The rate of calcification did not vary significantly after the perturbation (repeated measures ANOVA, P=0.3).

 $P=0.3\rangle$ . The calcification rate was significantly affected by the treatment (ANOVA, P<0.0001 for  $\rm FCO_2$  and P=0.3 for temperature). The significant interaction (P<0.001) between  $\rm pCO_2$  and temperature demonstrates that there was a response to a change in temperature but that it differs depending on the level of  $\rm pCO_2$ .

The rate of calcification declined immediately after the rise in pCO<sub>2</sub> and did not change afterwards, demonstrating that no acclimation process occurred. This is in agreement with previous studies: 
\* Marubini & Atkinson (1999) reported that the decrease of calcification is immediate in *Porites* 

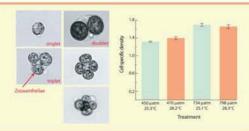
compressa.

\* Langdon et al. (2000) found that the response of the community calcification rate of the Biosphere 2 ocean is not different during short-term (days) and long-term (months) changes in

Calcification of colonies maintained at elevated temperature declined by 50% in response to increased pCO<sub>2</sub>. However, calcification was not affected by elevation of pCO<sub>2</sub> in colonies maintained at normal temperature. This is not in agreement with several papers that describe a negative relationship between calcification and CO<sub>2</sub> or a positive relationship by which the aragonite saturation state (Gattuso et al., 1998; Marubini & Atkinson, 1999; Langdon et al., 2000; Leclercq et al., 2002; Langdon et al., 2003; Marubini et al., 2003; Notweer, in some of these studies,  $\Omega_{\rm ENG}$  has not been changed by manipulating pCO<sub>2</sub> but by changing the Ca<sup>2+</sup>concentration (Gattuso et al., 1998), or by addition of acid Marubini & Thake 1999; Marubini et al., 2003) or sodium bicarbonate (Marubini & Atkinson, 1999). These results demonstrate that pCO<sub>2</sub> and temperature significantly interact to control calcification. The physiological basis of the different response at two temperatures does not result from an indirect effect of temperature on the seawater carbonate chemistry, Indeed, the change of pH and anagonite saturation state due to increased temperature was militar at both pCO<sub>2</sub> (Lght-1002 to -0.03;  $\Omega_{\rm EMB}^{\rm CMB}$ ). 81 on 25) and approximately 10 times lower than the changes resulting from increased pCO<sub>2</sub>.

# Cell Specific density (CSD)

3 nubbins from each treatment were used to determine the average number of zooxantheliae per animal cell (CSD). Consis were crushed with a hammer, placed in a 50 ml flask, and macerated by agitation (Muscatine & Cernichian, 1969). Approximately 330 host cells from each colony were observed and ranked according to the number of zooxantheliae (from one to eight) that each contained.



The CSD increased under 'high pCO<sub>2</sub>' (ANOVA, P < 0.001) without being affected by the change in temperature (ANOVA, P=0.4). There was a dominance of singlets over doublets or triplets under 'normal pCO<sub>2</sub>' (70% of singlets and 30% of doublets). The frequency distribution changed at elevated pCO<sub>2</sub> with 47% of singlets, 41% of doublets and 11% of animal cells containing more than 2 zooxanthellae.

Under normal pCO<sub>2</sub>, the CSD was identical at both temperature and equal to 1.4. The same value has been reported in the same species (Muscatine et al., 1998), and indicates that there is a dominiance of singlets. This seems to be the standard condition of the symbiosis. The CSD increased to 1.7 under elevated pCO<sub>2</sub> suggesting a higher rate of algal division compared to the division of animal cells. A change in CSD indicates a disruption of the balance between the growth rate of algal and animal cells.

These results are of major interest from a predictive point of view. Several studies investigated the physiological relationship between calcification and  $pCO_3$  or the aragonite saturation state. The consensus opinion is that calcification of tropical marine organisms and coral communities will decrease by an average 18-37% (Gattuso et al. 1999) between preindustrial time and the year 2100. However, none of these studies considered the effect of the forecast increase in temperature and its interaction with  $pCO_2$  on photosynthesis and calcification.

Our results demonstrate that the rate of calcification could decrease by 50% between the years 2000 and 2100. This temperature effect must be taken into consideration in subsequent investigations of future changes of coral physiology and reef metabolism. The present predictions must be re-evaluated as our results suggest that the decrease in the rate of calcification at the end of the century could be much higher than that forecast due to the synergistic effects of temperature and pCO<sub>2</sub>.

There is a pressing need to manipulate environmental parameters in concert in order to determine the response of coral calcification to global environmental changes.

### BIBLIOGRAPHY

Coles SL, Jokiel PL (1978) Synergistic effects of temperature, salinity and light on the hermatypic coral Montipora verrucoso. Mar Biol 48, 187-195
Device PS (1989) Short-term growth measurements of corals using an accurate buoyant weighing technique. Mar Biol 101, 389-395
Gattuso J-P. Allemand D. Franksipnoulle M (1999) Photosynthesis and calcification at cellular, organismal and community levels in coral reefs: A review on interactions and control by carbonate chemistry. Am zool 39, 160-183.
Gattuso J-P. Franksipoulle M. Bourge J. Romaine S. Buddeneier RW (1998) Effect of calcium carbonate saturation of seawater on coral calcification. Glob Planet Change 18, 37-46.
Goiran C. Al-Moghabi S. Allemand D. Jaseber J. (1996) inorganic carbon uptake for photosynthesis by the symbiotic coral/dinoflage/late association. J Exp Mar Biol Ecol 199, 207-225
Joseph P., Maragos J. Frantsisket J. (1978) Coral growth: buoyant weight technique. Stoddart DR, Johannes RE. Coral reef-research methods. 379-398. Paris, Unesco
Kajiwara K. Naga A. Lemos, S. Yakochi H. (1998) Examination of the effect of temperature. Ight intensity and zooxantheliae concentration on calcification and photosynthesis of scienactinian coral Acropora pulchro. J School Mar Sci Tech 40, 95-103

Langdon C, Takahashi T, Manubini F, Atkinson MJ, Sweeney C, Aceves H, Barnett H, Chipman D, Goddard J (2000) Effect of calcium carbonate saturation state on the rate of calcification of an experimental coral recf. Glob Biogocohem cycles 14,639-66 WS, Hammond DE, Gleon E, Fitzsimmons K, Nelson SG, Peng T-H, Hajdas I, Bonani G (2003) Effect Calcification O, Broecker WS, Hammond DE, Gleon E, Fitzsimmons K, Nelson SG, Peng T-H, Hajdas I, Bonani G (2003) Effect of elevated CO, on the community metabolism of an experimental coral recf. Glob Biogochem Cycles 17, 11.1–11.14 Leclerco, N. Gattuso J-P, Jaubert J (2000) CO<sub>2</sub> partial pressure controls the calcification rate of a coral community, Glob Change Biof 61, 1-6.

Leclerco, N. Gattuso J-P, Jaubert J (2002) Primary production, respiration, and calcification of a coral recf mesocosm underscripted CO<sub>2</sub> partial pressure is innot Change 27 SSE-SAG.

117-121
Marubini F, Ferrier-Pagès C, Cuif. JP (2003) Suppression of growth in scleractinian corals by decreasing ambient carbonate ion concentrations cross-family comparison. Proc Royal Society B 270, 179-184
Marubini F, Thake B (1999) Bicarbonate addition promotes coral growth. Limnod Oceanogr 44, 716-720
Muscatine L, Cenichiari E(1969) Assimilation of photosymthetic products of zooxanthellae by a reef coral. Biol Bull

Müscülere L. Cerricchien (1.1997) Abundundun du January (1997) (1